Acta Crystallographica Section E

## Structure Reports <br> Online

## 2'-(4-Methylbenzylidene)-5-phenyl-1H-pyrazole-3-carbohydrazide

ISSN 1600-5368

## Fang-Fang Jian,* Tong-Ling liang, Ying-Qi Qin, Huan-Qing Yu and Hai-Lian Xiao

New Materials and Function Coordination Chemistry Laboratory, Qingdao University of Science and Technology, Qingdao 266042,
People's Republic of China

Correspondence e-mail: ffj2003@163169.net

## Key indicators

Single-crystal X-ray study
$T=293 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.004 \AA$
$R$ factor $=0.051$
$w R$ factor $=0.165$
Data-to-parameter ratio $=14.5$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

[^0]The title compound, $\mathrm{C}_{18} \mathrm{H}_{16} \mathrm{~N}_{4} \mathrm{O}$, was prepared by the reaction of 5-phenyl- 1 H -pyrazole-3-carbohydrazide with 4-methylbenzaldehyde in refluxing ethanol. The structure exhibits weak intermolecular $\mathrm{N}-\mathrm{H} \cdots \mathrm{O}$ and $\mathrm{N}-\mathrm{H} \cdots \mathrm{N}$ hydrogenbonding interactions.

## Comment

Schiff bases have been used extensively as ligands in the field of coordination chemistry (Öğretir et al., 2006). As dinegatively charged ligands, Schiff bases show potential as antimicrobial and anticancer agents (Tarafder et al., 2000; Deschamps et al., 2003) and so have biochemical and pharmacological applications. In addition, the chemical behavior of metal complexes with Schiff base ligands has attracted much attention because of their catalytic activity in some industrial and biochemical processes ( Wu et al., 2006). The title compound, (I), was synthesized as part of our study of these ligands and we report here the crystal structure of (I).


In (I) (Fig. 1), the bond lengths and angles in $1 H$-pyrazole are in good agreement with those reported previously (Yang et al., 2003). The $\mathrm{C}=\mathrm{N}$ and $\mathrm{C}=\mathrm{O}$ bond lengths are comparable with those reported previously (Ali et al., 2005). Atoms N1, $\mathrm{N} 2, \mathrm{C} 7, \mathrm{C} 8, \mathrm{C} 9$ and C 6 lie in a plane ( $p 5$ ), with a maximum deviation of 0.0093 (3) $\AA$ for C7. Phenyl ring C1-C6 and atom C7 are coplanar ( $p 7$ ), with a maximum deviation of 0.0088 (3) $\AA$ for C7. Benzene ring C12-C17 is approximately coplanar with N 4 and $\mathrm{C} 18(p 9)$, with a maximum deviation of 0.0583 (3) A for N4. The dihedral angles made by $p 5$ with $p 7$ and $p 9$ are 23.12 (1) and $36.16(1)^{\circ}$, respectively. The dihedral angle between $p 7$ and $p 9$ is $58.99(1)^{\circ}$. There are some weak intermolecular $\mathrm{N}-\mathrm{H} \cdots \mathrm{O}$ and $\mathrm{N}-\mathrm{H} \cdots \mathrm{N}$ hydrogen bonding interactions (Table 2), which stabilize the structure of (I).

## Experimental

A mixture of 5-phenyl-1 H -pyrazole-3-carbohydrazide ( 0.01 mol ) with 4-methylbenzaldehyde ( 0.01 mol ) was stirred in refluxing ethanol $(30 \mathrm{ml})$ for 5 h to afford the title compound, (I) (yield: $2.77 \mathrm{~g}, 81 \%$ ). Single crystals suitable for X-ray measurements were obtained by recrystallization from dimethylformamide (DMF) at 309 K .

Received 31 August 2006
Accepted 2 September 2006

## Crystal data

| $\mathrm{C}_{18} \mathrm{H}_{16} \mathrm{~N}_{4} \mathrm{O}$ | $Z=8$ |
| :--- | :--- |
| $M_{r}=304.35$ | $D_{x}=1.166 \mathrm{Mg} \mathrm{m}^{-3}$ |
| Monoclinic, $C 2 / c$ | Mo $K \alpha$ radiation |
| $a=18.280(4) \AA$ | $\mu=0.08 \mathrm{~mm}^{-1}$ |
| $b=14.790(3) \AA$ | $T=293(2) \mathrm{K}$ |
| $c=13.364(3) \AA$ | Block, colorless |
| $\beta=106.27(3)^{\circ}$ | $0.18 \times 0.17 \times 0.15 \mathrm{~mm}$ |
| $V=3468.4(14) \AA^{\circ}$ |  |

## Data collection

Enraf-Nonius CAD-4
diffractometer
$\omega$ scans
Absorption correction: none
6290 measured reflections
3053 independent reflections

## Refinement

Refinement on $F^{2}$

$$
\begin{aligned}
& w= 1 /\left[\sigma^{2}\left(F_{\mathrm{o}}^{2}\right)+(0.0905 P)^{2}\right. \\
&+0.2393 P] \\
& \text { where } P=\left(F_{\mathrm{o}}{ }^{2}+2 F_{\mathrm{c}}^{2}\right) / 3 \\
&(\Delta / \sigma)_{\max }<0.001 \\
& \Delta \rho_{\max }=0.17 \mathrm{e}^{2} \AA^{-3} \\
& \Delta \rho_{\min }=-0.23 \mathrm{e} \AA^{-3}
\end{aligned}
$$

Extinction correction: SHELXL97
Extinction coefficient: 0.0061 (8)

Table 1
Selected geometric parameters $\left(\AA,{ }^{\circ}\right)$.

| $\mathrm{O} 1-\mathrm{C} 10$ | $1.239(2)$ | $\mathrm{N} 4-\mathrm{C} 11$ | $1.269(3)$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{N} 1-\mathrm{N} 2$ | $1.343(2)$ | $\mathrm{C} 6-\mathrm{C} 7$ | $1.462(3)$ |
|  |  |  |  |
| $\mathrm{N} 1-\mathrm{N} 2-\mathrm{C} 9$ | $104.44(17)$ | $\mathrm{C} 11-\mathrm{N} 4-\mathrm{N} 3$ | $115.60(19)$ |
| $\mathrm{C} 10-\mathrm{N} 3-\mathrm{H} 3 A$ | 120.4 | $\mathrm{O} 1-\mathrm{C} 10-\mathrm{N} 3$ | $123.8(2)$ |

Table 2
Hydrogen-bond geometry ( $\left(\AA^{\circ}{ }^{\circ}\right.$ ).

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :--- | :--- | :--- | :--- | :--- |
| $\mathrm{~N} 1-\mathrm{H} 1 A \cdots \mathrm{O} 1^{\mathrm{i}}$ | 0.86 | 2.10 | $2.813(2)$ | 140 |
| $\mathrm{~N} 3-\mathrm{H} 3 A \cdots \mathrm{~N} 2^{\mathrm{ii}}$ | 0.86 | 2.19 | $2.996(3)$ | 156 |



Figure 1
The molecular structure and atom-labeling scheme for (I), with displacement ellipsoids drawn at the $30 \%$ probability level.

H atoms were positioned geometrically and allowed to ride on their parent atoms, with $\mathrm{N}-\mathrm{H}=0.86 \AA$ and $\mathrm{C}-\mathrm{H}=0.93-0.96 \AA$, and with $U_{\text {iso }}(\mathrm{H})=1.2-1.5 U_{\text {eq }}$ of the parent atoms.

Data collection: CAD-4 Software (Enraf-Nonius, 1989); cell refinement: CAD-4 Software; data reduction: NRCVAX (Gabe et al., 1989); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL/PC (Sheldrick, 1990); software used to prepare material for publication: PARST (Nardelli, 1996) and WinGX (Farrugia, 1999).

The authors thank the Natural Science Foundation of Shandong Province (grant No. Y2005B04).

## References

Ali, H. M., Puvaneswary, S. \& Ng, S. W. (2005). Acta Cryst. E61, o3464-o3465. Deschamps, P., Kulkarni, P. P. \& Sarkar, B. (2003). Inorg. Chem. 42, 7366-7368. Enraf-Nonius (1989). CAD-4 Software. Version 5.0. Enraf-Nonius, Delft, The Netherlands.
Farrugia, L. J. (1999). J. Appl. Cryst. 32, 837-838.
Gabe, E. J., Le Page, Y., Charland, J.-P., Lee, F. L. \& White, P. S. (1989). J. Appl. Cryst. 22, 384-387.
Nardelli, M. (1996). J. Appl. Cryst. 29, 296-300.
Öğretir, C., Dal, H., Berber, H. \& Taktak, F. F. (2006). J. Chem. Eng. Data, 51, 46-50.
Sheldrick, G. M. (1990). SHELXTL/PC. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
Sheldrick, G. M. (1997). SHELXL97 and SHELXS97. University of Göttingen, Germany.
Tarafder, M. T. H., Ali, M. A., Wee, D. J., Azahari, K., Silong, S. \& Crouse, K. A. (2000). Transition Met. Chem. 25, 456-460.

Wu, L.-B., Hu, Z.-Q. \& Lai, G.-Q. (2006). Chin. J. Struct. Chem. 25, 567-571.
Yang, G. \& Raptis, R. G. (2003). J. Heterocycl. Chem. 32, 659-664.


[^0]:    (C) 2006 International Union of Crystallography All rights reserved

